Year: 5 Topic: Science Key concepts: Properties and changes of materials

Scientist spotlight: Rosalind Franklin

She was an English chemist and X-ray crystallographer whose work was central to the understanding of the molecular structures of DNA , RNA , viruses, coal, and graphite.



What you might already know:

- ⇒ about everyday materials and their properties and uses (Years 1 and 2)
- ⇒ about temperature and heating and cooling (Year 4)
- ⇒ about evaporation and condensation in the water cycle and the factors that affect evaporation (Year 4)

Our Learning Objectives:

- ⇒ To identify the properties of a range of materials and explain their uses.
- \Rightarrow To plan comparative or fair tests and then take accurate measurements and make accurate observations.
- ⇒ To explore making and separating mixtures.
- \Rightarrow To use relevant scientific language to explain their ideas.
- ⇒ To classify changes as reversible or irreversible.
- ⇒ To report and present findings from enquiries.

Fun facts!

- Geckskin[™] is a super strong adhesive that can hold up to 330kg on a smooth wall.
- A thread of spider silk breaks less easily than a thread of steel.
- ♦ Pearls will melt in vinegar.
- All materials on Earth are made from just 92 naturally occurring materials.

Scientific Core Knowledge:



The solid melts.

The liquid freezes.



liquid

to fall through the holes in

the sieve, separating them

from larger particles.

The gas condenses.

The liquid evaporates



a gas, leaving the solid

particles behind.

Sieving Filtering Evaporating Smaller materials are able The solid particles will The liquid changes into

get caught in the filter

paper but the liquid will

be able to get through.

Scientific Vocabulary:

Rigid: hard and fixed; not flexible.

Elastic: returns to original shape when force removed.

Flexible: easily bends; opposite of rigid and stiff.

Electrical conductor: material that electricity can flow through

Thermal conductor: material that allows heat to pass through it.

Solution: mixture of solid and liquid

Dissolve: when a solid mixes with liquid to make a solution.

Evaporate: heat liquid until it turns into gas.

Soluble: when something can dissolve.

Insoluble: when something can't dissolve.

Reversible change: one that can be undone

Irreversible change: one that cannot be undone

Misconceptions: You might think...

- that 'material' just means 'fabric'.
- that 'everyday materials' are single substances.
- that all liquids contain water.
- that dissolving means that the substance has disappeared.
- that melting and dissolving are the same thing.
- that rusting is a physical change.